

## PHOTOFERMENTATIVE PRODUCTION OF HYDROGEN BY *THIOPCAPSA ROSEOPERSICINA* FROM SIMPLE ORGANIC SUBSTRATES

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$H_2$  is an ideal, clean and potentially sustainable energy carrier for the future due to its large energy content per weight, abundance and non-polluting nature. The selection of optimal  $H_2$  production technology depends on the  $H_2$ -producing enzymes available. *Thiocapsa roseopersicina* contains a nitrogenase and several [NiFe] hydrogenases, which participate in  $H_2$  metabolism. In the present study,  $H_2$  production by the Hox1 soluble hydrogenase and the nitrogenase were investigated. The amount of  $H_2$  evolved by the nitrogenase enzyme was much higher than the amount produced by the Hox1 hydrogenase enzyme. By comparing the  $H_2$  production by nitrogenase from five short-chain organic acids (acetate, citrate, pyruvate, succinate, formate) the highest productivity of  $H_2$  (~3 times) was observed in the presence of 4 g/l pyruvate. In this case, the pyruvate consumption was 100%, the biomass growth was equal to that of the control, therefore the produced  $H_2$  derived from pyruvate.

**Keywords:** hydrogenase, nitrogenase, photofermentation, *Thiocapsa roseopersicina*, biohydrogen

### Introduction

Biohydrogen can be produced by anaerobic microorganisms and considered as a potential energy carrier of the future [1]. The anaerobic, non-photosynthetic bacteria decompose organic compounds (most frequently carbohydrates) into organic acids, carbon dioxide and  $H_2$ , but the subsequent anaerobic conversion of the organic acids is not feasible energetically in dark fermentation [2]. However, some photosynthetic bacteria manage to further exploit these organic acids, driven by solar energy due to the light-harvesting pigments found in the bacterial cell membrane. Being intimately linked to light energy, the process that makes possible the decomposition of organic acids into  $H_2$  and carbon dioxide is called photofermentation [3].

Hydrogenase enzymes play pivotal role in photofermentative  $H_2$  production as a biocatalyst of the reversible oxidation of molecular  $H_2$ .  $H_2$  is also generated by nitrogenases, which catalyze the reduction of molecular nitrogen into ammonia and that is accompanied by the reduction of protons to  $H_2$  [4, 5].

Photosynthetic bacteria have long been studied for their capacity to produce  $H_2$  [1]. The nitrogenase based  $H_2$  production in purple non-sulphur bacteria is the major field of research [6], while the study of nitrogenase mediated  $H_2$  production in purple sulphur bacteria seems to be a novel approach.

Our model organism, *Thiocapsa roseopersicina* BBS is an anaerobic, purple sulphur phototrophic bacterium which contains at least four distinct and active [NiFe] hydrogenases: two membrane-bound (HynSL, HupSL) and two soluble (Hox1EFUYH, Hox2FUYH) enzymes as well as a nitrogenase enzyme [7, 8]. In this study, the  $H_2$  productivity of the Hox1 soluble hydrogenase and the nitrogenase were compared. Moreover, the basic growth medium was supplemented with various organic carbon substrates in order to identify the ones the bacteria could use as electron source to produce  $H_2$ . The organic acids are utilized in diverse metabolic pathways. Electrons formed by biochemical reactions are transferred by cofactors as long as they get to the enzyme and become reduced to  $H_2$ .

### Materials and methods

#### *Bacterial strains and growth conditions*

Mutant strains of *T. roseopersicina* BBS (wild type) used in this study were GB1121 (*ΔhynSL*, *ΔhupSL*) and M539 (*ΔhypF*) [9] for Hox1-mediated and nitrogenase-based  $H_2$  production measurements, respectively. As negative control GB112131 (*ΔhynSL*, *ΔhupSL*, *Δhox1EFUYH*) and M539 (*ΔhypF*) under non-nitrogen fixing conditions were applied. HypF is an accessory

protein that is required for the biosynthesis of all active [NiFe] hydrogenases. All strains were grown anaerobically in liquid cultures with continuous illumination ( $35 \mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$  light intensity) at  $25^\circ\text{C}$  in Pfennig's mineral medium (2% NaCl, 0.1%  $\text{KH}_2\text{PO}_4$ , 0.1%  $\text{MgCl}_2$ , 0.1% KCl, 0.1%  $\text{NH}_4\text{Cl}$  as nitrogen source, 0.2%  $\text{NaHCO}_3$  as carbon source, 0.4%  $\text{Na}_2\text{S}_2\text{O}_3$ , 20  $\mu\text{g}/\mu\text{l}$  vitamin  $\text{B}_{12}$ , 3.3 mg/l Fe-EDTA, 2.975 mg/l Na<sub>2</sub>-EDTA, 0.3 mg/l  $\text{H}_3\text{BO}_3$ , 0.2 mg/l  $\text{CaCl}_2$ , 0.1 mg/l  $\text{ZnSO}_4$ , 0.03 mg/l  $\text{MnCl}_2$ , 0.03 mg/l  $\text{Na}_2\text{MoO}_4$ , 0.713 mg/l  $\text{NiCl}_2$ , 0.01 mg/l  $\text{CuCl}_2$ ). For nitrogen-fixing conditions, the  $\text{NH}_4\text{Cl}$  was omitted.

The growth media were supplemented with various carbon sources (acetate, citrate, formate, pyruvate, succinate) and tested at different initial concentrations (2 g/l, 4 g/l, 6 g/l).

#### *In vivo H<sub>2</sub> evolution activity measurements*

Cultures (20 ml and 60 ml) were grown in 27 ml and 100 ml hypovials; the gas phase was flushed with  $\text{N}_2$  after inoculation. The produced  $\text{H}_2$  was determined daily by gas chromatograph [9] (Agilent Technologies 6890N equipped with Molesieve 30 m x 0.53 mm x 25  $\mu\text{m}$  column and thermal conductivity detector; oven and detector temperature  $160^\circ\text{C}$ , mobil phase:  $\text{N}_2$ ).

#### *Organic-acid analysis*

In order to determine the residual organic acid concentrations in the culture medium, 1 ml of cell suspension was centrifuged at 13000 rpm for 10 min, and 50  $\mu\text{l}$  of the supernatant was analyzed by HPLC (Hitachi Elite, equipped with ICsep ICE-COREGEL 64H column and refractive index detector L2490) using the following parameters: solvent 0.1 N  $\text{H}_2\text{SO}_4$ , flow rate 0.8 ml/min, column temperature  $50^\circ\text{C}$ , detector temperature  $41^\circ\text{C}$ .

#### **Results and discussion**

##### *Biohydrogen production by Hox1 hydrogenase and nitrogenase enzymes*

The  $\text{H}_2$ -evolving enzymes used by most biohydrogen evolving systems are nitrogenases and hydrogenases. In order to compare the  $\text{H}_2$  productivity of the Hox1 [NiFe] hydrogenase and the nitrogenase enzymes in *T. roseopersicina*, two mutant strains were used.

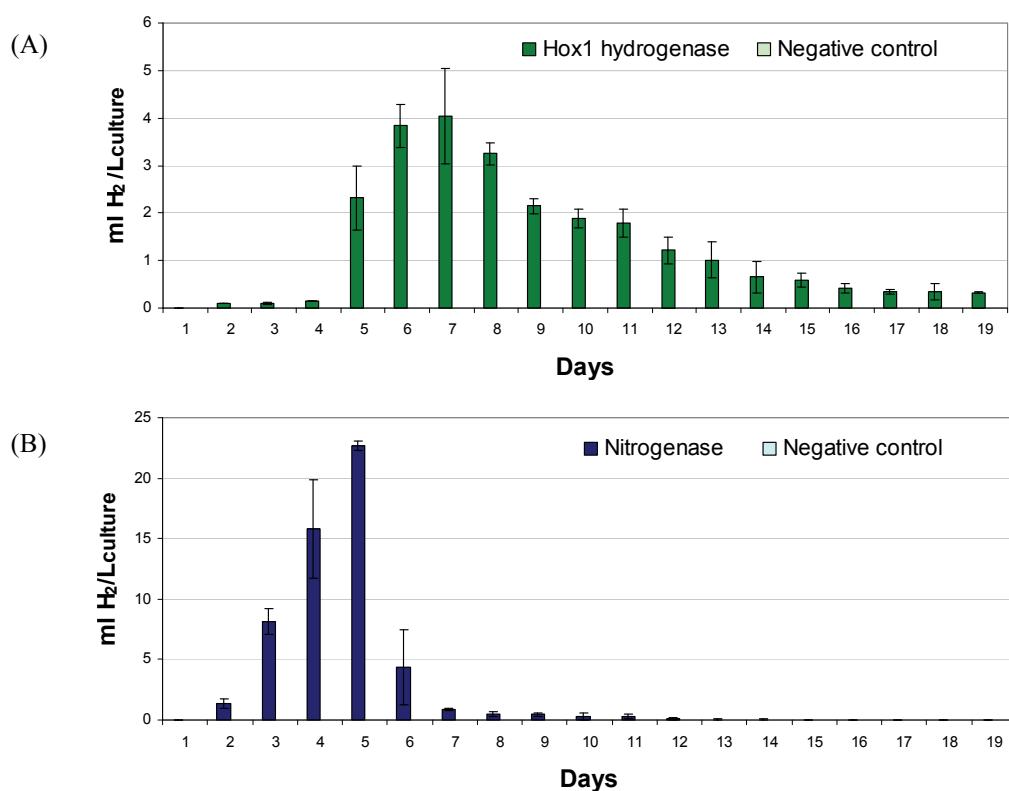


Figure 1: *In vivo* daily  $\text{H}_2$  production by Hox1 hydrogenase (A) and by nitrogenase (B)

The quantity of the  $\text{H}_2$  produced by Hox1 hydrogenase (GB1121 strain) and nitrogenase (M539 strain) enzymes was measured daily. It has to be noted, that although the GB1121 strain contains the genes encoding the Hox2 hydrogenase too, this enzyme does not evolve  $\text{H}_2$  in the circumstances studied [8]. It was observed that the amount of  $\text{H}_2$  produced by the nitrogenase enzyme under

nitrogen-fixing conditions was much higher than the amount produced by the Hox1 hydrogenase enzyme under nitrogenase repressed conditions (Fig. 1). However, almost no nitrogenase-based  $\text{H}_2$  production was observed after the 7<sup>th</sup> day, while the Hox1 hydrogenase-based  $\text{H}_2$  production lasted for two more weeks. Moreover, the daily removal of  $\text{H}_2$  from the headspace of the culture

which produces H<sub>2</sub> via Hox1 hydrogenase resulted in a relatively high H<sub>2</sub> production rate (1.018 ml H<sub>2</sub>/l culture/day). It has to be noted that, in all experiments, the strains used as negative control did not produce detectable amounts of H<sub>2</sub>.

### *Substrates for biohydrogen production*

Photosynthetic bacteria are able to produce H<sub>2</sub> from various reduced compounds; therefore, they are favourable substrates for biological H<sub>2</sub> production. The bacteria are capable of evolving H<sub>2</sub> from a wide range of organic acids, as well as from hydrogen sulphide, elemental sulphur or thiosulphate if CO<sub>2</sub> is supplied as carbon source. The conversion of different organic acids into H<sub>2</sub> – which would be advantageous for coupling clean, bioenergy production with organic waste-treatment – by different purple non-sulphur bacteria is well documented [10, 11], but little is known about the affinity of the purple sulphur bacteria to these organic acids. Therefore, by supplementing the growth medium (20 ml) with various organic acids, the conversion efficacies of these substrates by *T. roseopersicina* strains were tested. The experiments were repeated 3 times (except the asterisk labeled data in *Table 1*) using different initial concentrations of the substrates as listed in Table 1. The values given in Table 1 are the mean value of the data measured at day 9 of growth. The relative H<sub>2</sub> production refers to the H<sub>2</sub> produced by the same culture without supplementation. The H<sub>2</sub> produced by the cultures without supplementation in different runs are taken as 100%.

*Table 1:* Hydrogen production by hydrogenase (Hox1) and nitrogenase (N<sub>2</sub>ase)

Organic compound	Initial conc. (g/l)	H <sub>2</sub> production (ml H <sub>2</sub> /l culture)		Relative H <sub>2</sub> production (%)		Relative substrate consumption (%)	
		Hox1	N <sub>2</sub> ase	Hox1	N <sub>2</sub> ase	Hox1	N <sub>2</sub> ase
Acetate	2	0.51	110	16	90	74	74
	4	0.59	49	9	68	54	53
	6	0.46*	-	17*	-	46*	-
Citrate	2	2.27	110	65	94	-36	-30
	4	4.60	76	60	97	13	18
	6	0.90*	-	33*	-	44*	-
Pyruvate	2	2.48	219	63	<b>185</b>	100	99
	4	0.55	251	9	<b>307</b>	85	97
	6	0.59*	227*	22*	<b>211*</b>	97*	77*
Succinate	2	1.19	178	39	<b>151</b>	98	40
	4	2.31	214	30	<b>263</b>	71	32
	6	0.85*	226*	32*	<b>209*</b>	70*	31*
Formate	2	0.60	62	17	53	39	34
	4	0.81	33	14	45	45	47

### *The effect of pyruvate addition on the nitrogenase-based biohydrogen evolution*

In order to test the observation that in the case of nitrogenase, the extra H<sub>2</sub> was derived from the pyruvate, the same experiment was repeated with 2 g/l pyruvate in 60 ml culture volume. On day 13, when no H<sub>2</sub> production was observed, further pyruvate was added. During the

The data clearly show, that the Hox1 hydrogenase mediated H<sub>2</sub> production was negatively influenced by the organic acids present in the culture medium containing elevated amount of thiosulphate. However, at lower thiosulphate concentration, acetate driven hydrogen production could be observed (data not shown).

Supplementation of the nitrogen-fixing medium used for nitrogenase-based H<sub>2</sub> production with different organic acids resulted in higher H<sub>2</sub> production in case of the pyruvate and succinate. The increase of the initial concentration of pyruvate from 2 g/l to 4 g/l, increased the amount of H<sub>2</sub>, but further increase of the substrate concentration to 6 g/l decreased the H<sub>2</sub> production. Additionally, almost all of the added pyruvate was consumed in the first two cases (2 g/l and 4 g/l), while in the third case (6 g/l) the substrate consumption was only 77% in the first 9 days of the experiment.

Similar results were found in case of supplementing the growth medium with succinate with the difference that the volumes of H<sub>2</sub> produced were less, and the substrate consumption rates were lower, too.

In the course of experiments, the bacterial growth was measured beside the H<sub>2</sub> production. The application of 2 and 4 g/l pyruvate did not result in higher biomass growth (8.65 g cell/l culture and 8.4 g cell/l culture, respectively) compared to the pyruvate-free control (8.8 g cell/l culture); therefore it is likely that the extra H<sub>2</sub> derived from the pyruvate. Nevertheless, in the case of 6 g/l pyruvate not only the biomass concentration was increased (18.5 g cell/l culture), but also the specific H<sub>2</sub> production was enhanced.

experiments the total H<sub>2</sub> production, pyruvate concentration and the amount of biomass were measured daily. *Fig. 2A* illustrates the H<sub>2</sub> production with or without pyruvate and the variation of pyruvate concentration in time.

Visible improvement in the H<sub>2</sub> evolution was observed after supplementing with pyruvate, moreover the *in vivo* H<sub>2</sub> generation started at day 3, 2 days earlier than the control. This result is in agreement with the tendency of the bacterial growth (*Fig. 2B*).

The supplementation with pyruvate enhanced the bacterial growth, but at day 9 both culture reached the same level.

Further addition of pyruvate to the culture in stationary phase (day 13) resulted in a lower rate increase of H<sub>2</sub> production, which is caused by the repression of nitrogenase due to the production of organic nitrogen compounds.

Remarkably, a 2-fold difference in the amount of H<sub>2</sub> produced by nitrogenase from 2 g/l pyruvate in 20 ml and 60 ml cultures (220 ml H<sub>2</sub>/l culture and 420 ml H<sub>2</sub>/l culture, respectively) was observed. The pyruvate-free control did not show any difference in 20 ml and 60 ml culture (98 ml H<sub>2</sub>/l culture and 96 ml H<sub>2</sub>/l culture, respectively). These results indicate that accumulation of H<sub>2</sub> in the headspace may have an inhibitory effect on biohydrogen production, which has to be considered in scale-up experimental design.

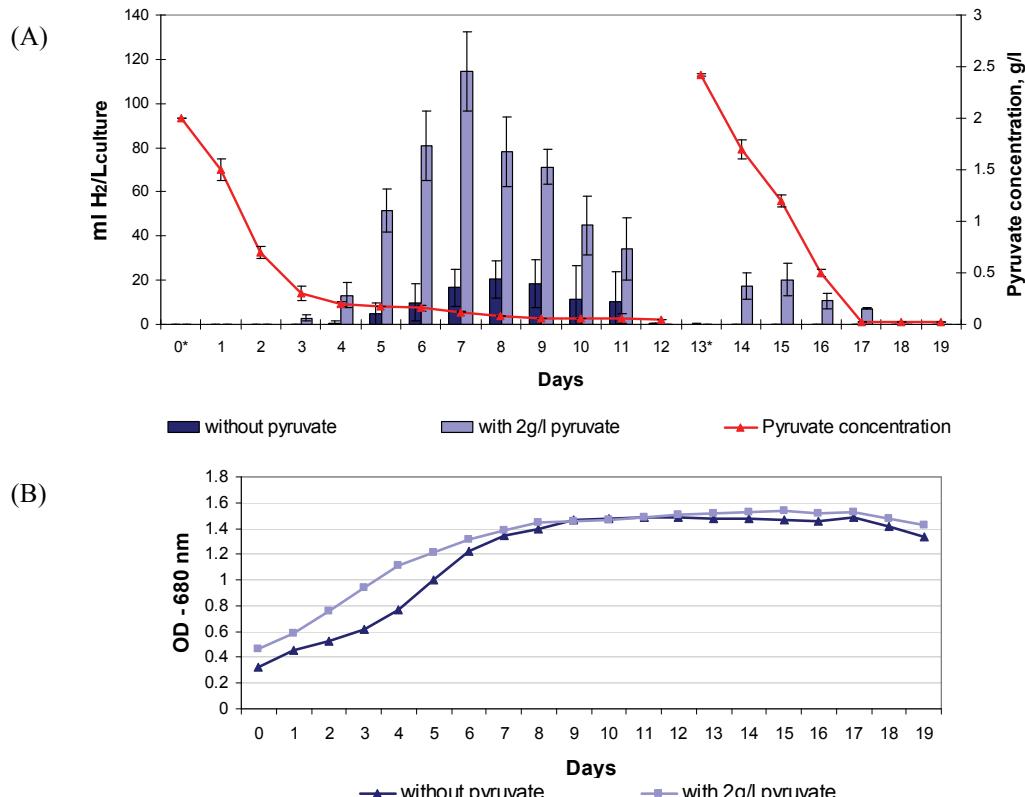


Figure 2: Effect of 2 g/l pyruvate on *in vivo* H<sub>2</sub> production by nitrogenase (A) and biomass growth (B)

In summary both nitrogenase and Hox1 hydrogenase can catalyze biohydrogen production from simple and inexpensive organic substrates in photoheterotrophic mode of growth. The overall H<sub>2</sub> yield generated by the nitrogenase system is higher, i.e., about 55 ml H<sub>2</sub>/l culture compared to the 25 ml H<sub>2</sub>/l culture for Hox1. Nitrogenase evolves H<sub>2</sub> intensively in short term while Hox1 [NiFe] hydrogenase is capable of sustained *in vivo* biohydrogen production, which may have important ramifications for large scale exploitation.

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#### REFERENCES

1. D. DAS, N. KHANNA, T. N. VEZIROĞLU: Chemical Industry and Chemical Engineering Quarterly, 14(2), 2008, 57–67.
2. P. C. HALLENBECK: International Journal of Hydrogen Energy, 34, 2009, 7379–7389.
3. A. ASADA, J. MIYAKE: Journal of Biosciences and Bioengineering, 88(1), 1999, 1–6.
4. P. TAMAGNINI, R. AXELSSON, P. LINDBERG, F. OXELFELT, R. WÜNSCHIERS, P. LINDBLAD: Microbiology and Molecular Biology Reviews, 66(1), 2002, 1–20.
5. K. L. KOVÁCS, G. MARÓTI, G. RÁKHÉLY: International Journal of Hydrogen Energy, 31, 2006, 1460–1468.

6. N. BASAK, D. DAS: *World Journal of Microbiology and Biotechnology*, 23, 2007, 31–42.
7. K. L. KOVÁCS, Á. T. KOVÁCS, G. MARÓTI, L. S. MÉSZÁROS, J. BALOGH, D. LATINOVICS, A. FÜLÖP, R. DÁVID, E. DOROGHÁZI, G. RÁKHELY: *Proceedings of 7th International Conference on Hydrogenases*, 2004, 61–63.
8. J. MARÓTI, A. FARKAS, I. K. NAGY, G. MARÓTI, É. KONDOROSI, G. RÁKHELY, K. L. KOVÁCS: *Applied and Environmental Microbiology*, 76(15), 2010, 5113–5123.
9. B. FODOR, G. RÁKHELY, Á. T. KOVÁCS, K. L. KOVÁCS: *Applied and Environmental Microbiology*, 67, 2001, 2476–2483.
10. M. J. BARBOSA, J. M. S. ROCHA, J. TRAMPER, R. H. WIJFFELS: *Journal of Biotechnology*, 85, 2001, 25–33.
11. H. KOKU, İ. EROĞLU, U. GÜNDÜZ, M. YÜCEL, L. TÜRKER: *International Journal of Hydrogen Energy*, 27, 2002, 1315-1329.